# organic compounds

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# [<sup>2</sup>H<sub>3</sub>]Sarcosine

#### Monika Trzebiatowska-Gusowska<sup>a</sup> and Anna Gagor<sup>b\*</sup>

<sup>a</sup>Wroclaw University of Technology, Wybrzeze Wyspianskiego 27, 50-370 Wroclaw, Poland, and <sup>b</sup>W. Trzebiatowski Institute of Low Temperature and Structure Research, Polish Academy of Sciences, Okólna str. 2, PO Box 1410, 50-950 Wrocław, Poland

Correspondence e-mail: a.gagor@int.pan.wroc.pl

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Key indicators: single-crystal X-ray study; T = 286 K; mean  $\sigma$ (C–C) = 0.002 Å; R factor = 0.028; wR factor = 0.075; data-to-parameter ratio = 9.5.

The crystal structure of  $[^{2}H_{3}]$ sarcosine or methyl $[^{2}H_{3}]$ ammonioacetate,  $CD_{3}NH_{2}^{+}CH_{2}COO^{-}$  or  $C_{3}H_{4}D_{3}NO_{2}$ , was solved at room temperature. It is isostructural with nondeuterated sarcosine. The amino acid molecules exist as zwitterions. Each molecule is connected to four neighbouring molecules through  $N-H\cdots O$  hydrogen bonds.

#### **Related literature**

Undeuterated sarcosine was reported by Mostad & Natarajan (1989).



#### **Experimental**

Crystal data	
$C_3H_4D_3NO_2$	a = 6.7998 (13) Å
$M_r = 92.11$	b = 7.9208 (16) Å
Orthorhombic, $P2_12_12_1$	c = 8.5874 (18) Å

 $V = 462.52 (16) \text{ Å}^3$ Z = 4Mo *K*\alpha radiation

#### Data collection

Kuma KM-4 diffractometer with	734 independent reflections
CCD area detector	608 reflections with $I > 2\sigma(I)$
Absorption correction: none 5729 measured reflections	$R_{\rm int} = 0.030$

 $\mu = 0.11 \text{ mm}^{-1}$ T = 286 (2) K

 $0.15 \times 0.1 \times 0.1 \text{ mm}$ 

#### Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.028$ 77 parameters $wR(F^2) = 0.075$ Only H-atom coordinates refinedS = 0.96 $\Delta \rho_{max} = 0.12$  e Å<sup>-3</sup>734 reflections $\Delta \rho_{min} = -0.13$  e Å<sup>-3</sup>

#### Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
	0.867 (18) 0.905 (18)	1.971 (18) 1.860 (19)	2.7886 (16) 2.7599 (17)	156.9 (16) 172.4 (15)
$C3 - D5 \cdots O2^{iii}$	1.00 (2)	2.35 (2)	3.355 (2)	179 (2)
Symmetry codes: (i	$(-x, y - \frac{1}{2}, -z +$	$\frac{1}{2}$ ; (ii) $-x + \frac{1}{2}, -y$	$z = \frac{1}{2}$ ; (iii) $x = \frac{1}{2}$ , -	$-y - \frac{1}{2}, -z.$

Data collection: *CrysAlis* (Mayer, 2006); cell refinement: *CrysAlis*; data reduction: *CrysAlis*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *DIAMOND* (Brandenburg, 2005); software used to prepare material for publication: *publCIF* (Westrip, 2007).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: ZL2085).

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supplementary materials

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# [<sup>2</sup>H<sub>3</sub>]Sarcosine

## M. Trzebiatowska-Gusowska and A. Gagor

## Comment

We determined the crystal structure of sarcosine-d3 (methyl-d3). It is isostructural with sarcosine  $-P2_12_12_1$  (Mostad & Natarajan, 1989). The sarcosine molecules are in the form of zwitterions creating helices down the a, b and c axes through the network of N—H···O hydrogen bonds. There are also weak C(3)—D(5)···O(2) interactions with C(3)···O(2) = 3.355 (1) Å and a C(3)—D(5)···O(2) angle of 178.60 (1) °. Figure 1 shows the molecule of the sarcosine. Figure 2 shows the packing of sarcosine-d3 molecules in the crystal as seen down the *b* axis. Vibrational spectroscopy measurements corroborate the presence of the deuterated methyl group (–CD<sub>3</sub>).

## Experimental

The single crystals of sarcosine-d3 were obtained from Aldrich (99 atom % D).

### Refinement

Positions of all hydrogen and deutrium atoms were taken from difference Fourier maps. They were refiened with isotropic displacement parameters correlated with the anisotropic displacement parameters of the atoms to which they were bonded  $[U_{iso}(H) = 1.2U_{eq}(C)]$ . Friedel pairs were merged prior to refinement

### Figures



Fig. 1. The molecule of sarcosine-d3.

Fig. 2. The packing of sarcosine-d3 molecules in the crystal as seen down the b axis. The dashed lines represent the hydrogen bonds.

## methyl[<sup>2</sup>H<sub>3</sub>]ammoniobenzoate

Crystal data
C <sub>3</sub> H <sub>4</sub> D <sub>3</sub> NO <sub>2</sub>
$M_r = 92.11$
Orthorhombic, $P2_12_12_1$
Hall symbol: P 2ac 2ab
<i>a</i> = 6.7998 (13) Å
<i>b</i> = 7.9208 (16) Å

 $F_{000} = 192$   $D_x = 1.323 \text{ Mg m}^{-3}$ Mo K\alpha radiation  $\lambda = 0.71073 \text{ Å}$ Cell parameters from 734 reflections  $\theta = 3.8-29.4^{\circ}$   $\mu = 0.11 \text{ mm}^{-1}$  T = 286 (2) KPrism, colourless  $0.15 \times 0.1 \times 0.1 \text{ mm}$ 

#### Data collection

Z = 4

c = 8.5874 (18) Å $V = 462.52 (16) \text{ Å}^3$ 

Kuma KM-4 diffractometer with CCD area detector	734 independent reflections
Radiation source: fine-focus sealed tube	608 reflections with $I > 2\sigma(I)$
Monochromator: graphite	$R_{\rm int} = 0.030$
Detector resolution: 1024x1024 with blocks 2x2,	$\theta = -20 1^{\circ}$
33.133pixel/mm pixels mm <sup>-1</sup>	$0_{max} = 29.4$
T = 286(2)  K	$\theta_{\min} = 3.8^{\circ}$
ω–scan	$h = -8 \rightarrow 9$
Absorption correction: none	$k = -9 \rightarrow 10$
5729 measured reflections	$l = -11 \rightarrow 11$

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_0^2) + (0.053P)^2]$ where $P = (F_0^2 + 2F_c^2)/3$
Least-squares matrix: full	$(\Delta/\sigma)_{\rm max} = 0.002$
$R[F^2 > 2\sigma(F^2)] = 0.028$	$\Delta \rho_{max} = 0.12 \text{ e } \text{\AA}^{-3}$
$wR(F^2) = 0.075$	$\Delta \rho_{\rm min} = -0.13 \text{ e } \text{\AA}^{-3}$
<i>S</i> = 0.96	Extinction correction: SHELXL97, Fc <sup>*</sup> =kFc[1+0.001xFc <sup>2</sup> $\lambda^{3}$ /sin(20)] <sup>-1/4</sup>
734 reflections	Extinction coefficient: 0.077 (15)
77 parameters	
Primary atom site location: structure-invariant direct methods	
Secondary atom site location: difference Fourier map	

Only H-atom coordinates refined

### Special details

**Geometry**. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement**. Refinement of  $F^2$  against ALL reflections. The weighted *R*-factor *wR* and goodness of fit S are based on  $F^2$ , conventional *R*-factors *R* are based on F, with F set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2 \operatorname{sigma}(F^2)$  is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on  $F^2$  are statistically about twice as large as those based on F, and R– factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(A^2)$ 

	x	У	Ζ	$U_{\rm iso}*/U_{\rm eq}$
01	0.21502 (16)	0.18017 (13)	0.32862 (12)	0.0404 (3)
O2	0.2292 (2)	-0.09212 (14)	0.26347 (14)	0.0525 (4)
N1	-0.01173 (18)	-0.06244 (15)	0.01915 (14)	0.0293 (3)
H4	-0.046 (3)	-0.141 (2)	0.084 (2)	0.035*
H3	0.092 (2)	-0.102 (2)	-0.036 (2)	0.035*
C1	0.1734 (2)	0.05453 (18)	0.24312 (17)	0.0314 (3)
C2	0.0441 (2)	0.09257 (18)	0.10361 (17)	0.0320 (3)
H2	-0.074 (3)	0.146 (2)	0.135 (2)	0.038*
H1	0.104 (2)	0.163 (2)	0.030 (2)	0.038*
C3	-0.1714 (3)	-0.0354 (2)	-0.0949 (2)	0.0467 (4)
D5	-0.199 (3)	-0.146 (3)	-0.147 (3)	0.056*
D6	-0.134 (3)	0.056 (3)	-0.170 (2)	0.056*
D7	-0.285 (3)	0.003 (2)	-0.043 (2)	0.056*

Atomic displacement parameters  $(Å^2)$ 

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
01	0.0485 (6)	0.0358 (6)	0.0370 (6)	0.0047 (5)	-0.0085 (5)	-0.0113 (5)
02	0.0736 (8)	0.0307 (6)	0.0532 (7)	0.0042 (5)	-0.0275 (6)	-0.0011 (5)
N1	0.0361 (6)	0.0255 (6)	0.0263 (6)	-0.0012 (5)	-0.0016 (5)	0.0013 (5)
C1	0.0352 (6)	0.0307 (7)	0.0282 (7)	-0.0019 (6)	0.0001 (6)	-0.0012 (6)
C2	0.0417 (7)	0.0256 (7)	0.0289 (7)	0.0003 (6)	0.0001 (6)	-0.0011 (6)
C3	0.0532 (10)	0.0388 (9)	0.0481 (10)	-0.0016 (8)	-0.0219 (8)	0.0011 (8)

*Geometric parameters (Å, °)* 

O1—C1	1.2687 (17)	C1—C2	1.516 (2)
O2—C1	1.2343 (18)	С2—Н2	0.949 (18)
N1—C2	1.4757 (18)	C2—H1	0.931 (18)
N1—C3	1.4782 (19)	C3—D5	1.00 (2)
N1—H4	0.867 (18)	C3—D6	1.00 (2)
N1—H3	0.905 (18)	C3—D7	0.94 (2)

# supplementary materials

C2—N1—C3	113.22 (12)	C1—C2—H2	111.1 (11)
C2—N1—H4	110.9 (11)	N1—C2—H1	106.0 (11)
C3—N1—H4	109.1 (11)	С1—С2—Н1	113.5 (10)
C2—N1—H3	110.2 (11)	H2—C2—H1	106.9 (15)
C3—N1—H3	106.2 (10)	D5-C3-D6	113.5 (17)
H4—N1—H3	107.0 (16)	D5—C3—D7	109.8 (16)
O2—C1—O1	126.00 (14)	D6—C3—D7	105.7 (18)
O2—C1—C2	118.49 (12)	D5—C3—N1	107.8 (13)
O1—C1—C2	115.51 (12)	D6—C3—N1	110.2 (12)
N1—C2—C1	111.84 (11)	D7—C3—N1	109.8 (13)
N1—C2—H2	107.0 (11)		
C1—C2—N1—C3	167.20 (14)		

## *Hydrogen-bond geometry (Å, °)*

D—H···A	<i>D</i> —Н	H···A	$D \cdots A$	D—H··· $A$	
N1—H4···O1 <sup>i</sup>	0.867 (18)	1.971 (18)	2.7886 (16)	156.9 (16)	
N1—H3···O1 <sup>ii</sup>	0.905 (18)	1.860 (19)	2.7599 (17)	172.4 (15)	
C3—D5···O2 <sup>iii</sup>	1.00 (2)	2.35 (2)	3.355 (2)	178.6 (17)	
Symmetry codes: (i) $-x$ , $y-1/2$ , $-z+1/2$ ; (ii) $-x+1/2$ , $-y$ , $z-1/2$ ; (iii) $x-1/2$ , $-y-1/2$ , $-z$ .					





